



as electrocatalysts for syngas production



<u>Hannah Rabl^a, Stephen N. Myakala^a, Jakob Rath^a, Bernhard Fickl^a, Jasmin S. Schubert^a, Dogukan H.</u> Apaydin^a, Dominik Eder^a

^a Institute of Materials Chemistry, Vienna University of Technology, Vienna, Austria

Check out this publication here

1. Introduction

1.1 Motivation

Electrocatalytic CO₂ reduction in aqueous media is an efficient method for reducing the CO₂ in our atmosphere and converting it into a more valuable feedstock such as for example syngas, a mixture of H_2 and CO. It further has the great advantage of being operated at room temperatures and ambient pressures. One major challenge in the field however is the search for selective, stable and efficient catalysts.

1.2 Metal organic chalcogenolate assemblies (MOCHAs)

Metal organic chalcogenolate assemblies (MOCHAs) such as $[AgSePh]_{\infty}$ also referred to as mithrene or $[AgSPh]_{\infty}$ (thiorene) are organicinorganic hybrid materials with outstanding optoelectronic properties^{1,2}. They were recently reported to work as X-ray³ and Pb²⁺ sensors⁴. This work has shown their potential as suitable electrocatalysts for syngas formation⁵.

2.1 Microwave synthesis of MOCHAs







An H-cell setup filled with 0.5M KHCO₃ on

both sides was chosen. The working side

compartment was saturated with CO₂ prior

to electrolysis. The MOCHA/CP electrodes

were used as working electrodes. An

Ag/AgCl/3M KCl was used as reference

electrode and Pt-foil was used as counter





3. Electrocatalytic Syngas formation

4. Stability studies

3.1 H-cell setups



3.2. Syngas formation

Bare CP does not lead to any CO production. Thiorene ($[AgSPh]_{\infty}$) outperforms mithrene $([AgSePh]_{\infty})$ by approx. 17 µmol h⁻¹ regarding CO production. The syngas ratio (H₂:CO) can be tuned from 1 to 10 depending on the applied potential.

electrode.

4.1 Recyclability of $[AgSePh]_{\infty}$ upon electrolysis

2. Synthesis



Using the same [AgSePh]_∞/CP electrode for four consecutive electrolysis runs revealed recyclability of the MOCHA catalyst. A stable CO:H₂ ratio was obtained for all four runs.



4.2 Stability of $[AgSePh]_{\infty}$ upon various conditions



Stability tests reveal structural (pXRD) and morphological (SEM) stability of $[AgSePh]_{\infty}$ under prolonged exposure (>1 week) to a wide pH range and various organic solvents.



6. Take-home messages

- A facile, fast and upscalable synthesis method for metal organic chalcogenolate assemblies was developed.
- First application of the MOCHAs [AgSePh] $_{\infty}$ (mithrene) and [AgSPh] $_{\infty}$ (thiorene) for electrocatlytic syngas formation was shown.



- The ratio of syngas (CO: H_2) can be tuned by the applied potential. ullet
- Recyclability and stability of $[AgSePh]_{\infty}$ was shown upon 4 consecutive electrolysis runs. \bullet
- MOCHAs are stable over a wide pH and temperature range and in various organic solvents. \bullet

Contact	References	Acknowledgements
hannah.rabl@tuwien.ac.at	 [1] Schriber, E. A. <i>et al.</i> Mithrene Is a Self-Assembling Robustly Blue Luminescent Metal–Organic Chalcogenolate Assembly for 2D Optoelectronic Applications. <i>ACS Appl. Nano Mater.</i> 1, 3498–3508 (2018). [2] Schriber, E. A. <i>et al.</i> Chemical crystallography by serial femtosecond X-ray diffraction. <i>Nature</i> 601, 360–365 (2022). [3] Fratelli, I. <i>et al.</i> Layered metal-organic chalcogenide thin films for flexible and large-area X-ray direct detection. <i>Front. Phys.</i> 11, 1325164 (2023). 	USEM ************************************
Find us on Find us on State of the search Group Find us on For Business of the search Group Find us on For Business of the search Group For Busi	 [4] Ye, C. <i>et al.</i> Dual-Emission 2D Blue Luminescent Organic Silver Chalcogenide for Highly Selective Pb²⁺ Detection in an Aqueous Medium. <i>Inorg. Chem.</i> 62, 2334–2341 (2023). [5] Rabl, H. <i>et al.</i> Microwave-assisted synthesis of metal-organic chalcogenolate assemblies as electrocatalysts for syngas production. <i>Commun Chem</i> 6, 43 (2023). 	TIU EDER GROUP